Characteristics and Mechanism of Salinized Triethanolamine as Cement Grinding Aids

Changsen ZHANG  
School of Materials Engineering Yancheng Institute of Technology  
Yancheng, P.R. China  
E-mail: zcsen1@163.com

Jianli ZHANG  
School of Materials Engineering Yancheng Institute of Technology  
School of Materials Science and Engineering Changzhou University  
Changzhou, P.R. China  
E-mail: kuaile_zjl@163.com

Zhenzhe FENG  
School of Materials Science and Engineering Jiangsu University  
Zhenjiang, P.R. China  
E-mail: fzzhe1@163.com

Baogui ZHU  
School of Materials Science and Engineering Jiangsu University  
Zhenjiang, P.R. China  
E-mail: 18262383300@163.com

Yang LI  
School of Materials Science and Engineering Changzhou University  
Changzhou, P.R. China  
E-mail: 18262383539@163.com

Abstract—Effects of salinized triethanolamine as cement grinding aids on particles characteristics and mechanical property of cement were studied, and its reaction mechanism was analyzed by FT-IR, SEM and XRD. According to the results, when the amount of triethanolamine salinization(DGA) was 0.03%, the sieve residue of 45 μm was lowered by 47.2%, specific surface area was increased by 45 m²/kg, the particle size distribution of cement which ranges from 3 to 32 μm increased 10.6%. The compressive strength of cement in 3, 7 and 28 d were increased by 4.9, 5.0 and 3.5 MPa, respectively. Enhancement of DGA on cement mainly lied in that it could promote or induce hydration reaction of cement mineral with gypsum and water, which accelerated formation of hydration products, and then improved the structure and morphology of cement hydration products, thus the uniformity and compactness of hydration products structure increased.

Keywords—salinized triethanolamine; grinding aids; particles properties; strength; cement

I. INTRODUCTION

Cement world production currently increases up to approximately 4 billion tons/year. The electrical energy consumed in cement production is 110 KW/tone and about 30% of which is used for raw material preparation and about of 40% for final cement clinker grinding[1-3]. This leads to important cost increase of the final product, especially under the present energy scarce environment[4].

During comminution of clinkers, highly reactive positive and negative charges are created on the newly fractured surfaces. As a result, the agglomeration phenomenon of solid particles occurred due to Van der Waals force and electrostatic attraction, which results in the formation of coatings in the grinding media and reduces the grinding efficiency[5]. In order to improve the energy efficiency of cement production process, one of successful directions is extensive use of grinding aid in comminution process. Because of highly organic polar nature, grinding aid is generally absorbed on particle surface through physical or chemical interaction with cement salts[6]. Due to particle surface state transformation, new charges are neutralized and attractive surface forces are screened, leading to lower grain surface hardness and better dispersion ability of particles.

Although there are a variety of grinding aids in the cement industry, the triethanolamine (TEOA) is extensive used at domestic and overseas. TEOA has a good comprehensive performance that it can improve the effect of grinding, promote the hydration process and enhance the early strength of cement[7-8]. However, TEOA also has limitations and shortcomings that application performance of TEOA is sensitive to its dosage and expensive costing[9-10]. So, with the development trend of cement industry, it is necessary to modify TEOA to improve performance and lower the cost.

During the study on TEOA modification by organic carboxylic acid, we found incidentally that by proper chemical modification TEOA had excellent grinding performance. The overall goal is to modify TEOA and
research its effect on grindability and cement performance including standard consistency and compressive strengths. To the best of our knowledge, this is the seldom report about the usage of organic carboxylic acid to salinize TEOA as cement grinding aids. As known, the nitrogen atoms of TEOA structure has the lone pair electrons, so, TEOA has the characteristics of weak alkaline. And then, TEOA can react with organic acid to form a salt improving its performance[11]. Therefore, this paper mainly focused on the grinding performance and effect of DGA on cement properties.

II. EXPERIMENTAL

A. Materials

Triethanolamine, Organic carboxylic acid and sodium hydroxide were all analytical grades, and purchased from Tianjin Chemical Reagent Co. Ltd. (China). Deionized water was used throughout the following work. All other reagents were analytic grade without further purification. Cement clinker and gypsum was supplied by Jiangsu baling, conch cement Co. Ltd

<table>
<thead>
<tr>
<th>%</th>
<th>SiO₂</th>
<th>CaO</th>
<th>Al₂O₃</th>
<th>Fe₂O₃</th>
<th>MgO</th>
<th>SO₃</th>
<th>LOI</th>
</tr>
</thead>
<tbody>
<tr>
<td>Clinker</td>
<td>21.8</td>
<td>65.8</td>
<td>5.4</td>
<td>3.3</td>
<td>1.7</td>
<td>0.4</td>
<td>0.4</td>
</tr>
<tr>
<td>Gypsum</td>
<td>3.3</td>
<td>37.6</td>
<td>1.3</td>
<td>0.3</td>
<td>1.3</td>
<td>36.2</td>
<td>15.8</td>
</tr>
</tbody>
</table>

(China). The chemical composition of the clinker and gypsum used are reported in Tab. 1.

Triethanolamine modifier (DGA): it was synthesized by TEOA, organic carboxylic acid through specific synthesis process of salinization reaction. And then the reactant is adjusted by sodium hydroxide to neutral. DGA can dissolve with water.

The infrared spectra(IR) of TEOA and DGA are shown in Tab.2. It can be seen from Fig. 1 that the stretching vibration band of -OH at 3355 cm⁻¹ and the stretching vibration band of -CH₃, -CH₂- at 2919 cm⁻¹ of DGA were nearly consistent with the stretching vibration of TEOA. At the same time, FT-IR of DGA had the stretching vibration band of C-O nearby at 1076 cm⁻¹, the stretching vibration band of C-N at 1031 cm⁻¹ and the deformation vibration band of -CH₃, -CH₂ at 1405 cm⁻¹. The FT-IR of DGA had the carboxylic acid stretching vibration band of -C=O at 1724 cm⁻¹, the stretching vibration band of N-H at 3151 cm⁻¹ and the deformation vibration band of N-H at 1563 cm⁻¹, while there weren’t these stretching vibration bands in FT-IR of TEOA. These changes showed that there were -C=O, N-H functional groups in the molecular structure of DGA, but not in TEOA.

![Figure 1. FT-IR spectra of TEOA and DGA](image_url)

B. Methods

a) Grinding experimental methods:

The cement clinker was crushed by PE60 mm × 100 mm jaw crushe, and then clinker with the same size was selected.

The selected cement clinker and dihydrate gypsum (the mass ratio of 95:5) were put into a QM-3SP2 planetary ball mill for grinding. The TEOA and DGA were respectively added into planetary ball mill in different dosages: 0.01%, 0.03%, 0.05%, 0.07%, 0.1% (weight percent of solid content to clinker and gypsum). Grinding time is 20 min.

b) Test methods:

IR was tested and analyzed by the Fourier transform infrared spectroscopy according to the Chinese National Standard GB/T6040-2002; The testing method for fineness of cement—Sieving method (GB/T1345-2005); The testing method for specific surface area of cement—Blaine method (GB/T8074-2008); The testing method for particle size distribution of cement—Laser method (JC/T721-2006); The testing method of strength for cement mortar—ISO method (GB/T17671-1999); The morphology, mineral phase and hydration degree of cement paste hydration products in different age were analyzed with scanning electron microscopy and X-ray diffraction, respectively.

III. RESULTS AND DISCUSSION

A. FT-IR Analysis

The FT-IR spectra of TEOA and DGA are shown in Fig.1. It can be seen from Fig.1 that the stretching vibration band of -OH at 3355 cm⁻¹ and the stretching vibration band of -CH₃, -CH₂- at 2919 cm⁻¹ of DGA were nearly consistent with the stretching vibration of TEOA. At the same time, FT-IR of DGA had the stretching vibration band of C-O nearby at 1076 cm⁻¹, the stretching vibration band of C-N at 1031 cm⁻¹ and the deformation vibration band of -CH₃, -CH₂ at 1405 cm⁻¹. The FT-IR of DGA had the carboxylic acid stretching vibration band of -C=O at 1724 cm⁻¹, the stretching vibration band of N-H at 3151 cm⁻¹ and the deformation vibration band of N-H at 1563 cm⁻¹, while there weren’t these stretching vibration bands in FT-IR of TEOA. These changes showed that there were -C=O, N-H functional groups in the molecular structure of DGA, but not in TEOA.

B. Effect of DGA and TEOA on Particles Characteristics of Cement

Particles characteristics of cement with different dosages of TEOA and DGA are shown in Tab.2. It can be seen from Tab.2 that the fineness of cement with TEOA and DGA are significantly decreased and the specific surface area is apparently increased compared to blank sample. It is shown that TEOA and DGA can significantly reduce the content of coarse particles and increase a few content of fine particles, and the effect of DGA is better than TEOA.

The particle size distribution is the percentage of different diameter particle in all cement particles which is the most important parameter for Portland cement, and directly affects the chemical and mechanical properties of the cement. Researchers have already indicated that the particle of 3~32 μm play a major role in increasing the strength of cement[12,13].
When the dosage of TEOA and DGA are 0.03%, the specific surface area of TEOA and DGA are separately increased by 31 m²/kg, 45 m²/kg and 45 μm sieve residue are decreased respectively by 4.1% and 5.0%.

In all cases the adding of DGA results in higher content of particles (3~32 μm) in comparison with the blank of non-additive cement. And the effect of DGA on the content of 3~32 μm particles is 5.9% more than that of blank sample when the dosage of DGA is 0.03%, which also is 1.9% more than the cement with the same dosage of TEOA.

The results showed that DGA was effective in improvement of cement grindability. After joining the DGA, the cement was introduced the reactive groups. These reactive groups reduced the particle's surface energy, reduced the stress needed for the crack propagation and neutralize the unsaturated electricity of surface, and then improved effective content of particle.

**TABLE II. FINENESS AND PARTICLE SIZE DISTRIBUTION OF CEMENT WITH DIFFERENT DOSAGES OF TEOA, DGA**

<table>
<thead>
<tr>
<th>Sample</th>
<th>Dosage (%)</th>
<th>Specific surface (m²/kg)</th>
<th>45 μm sieve residue (%)</th>
<th>Particle size distribution(%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Blank</td>
<td>0</td>
<td>340</td>
<td>10.6</td>
<td>≤3 μm</td>
</tr>
<tr>
<td></td>
<td>0.03</td>
<td>371</td>
<td>6.5</td>
<td>7.9</td>
</tr>
<tr>
<td>TEOA</td>
<td>0.05</td>
<td>369</td>
<td>7.1</td>
<td>10.8</td>
</tr>
<tr>
<td>DGA</td>
<td>0.05</td>
<td>382</td>
<td>5.6</td>
<td>10.5</td>
</tr>
</tbody>
</table>

**TABLE III. PHYSICAL AND MECHANICAL PROPERTIES OF CEMENT WITH DIFFERENT TYPES AND DOSAGES OF GRINDING AIDS**

<table>
<thead>
<tr>
<th>Sample</th>
<th>Dosage (%)</th>
<th>Soundness</th>
<th>Compressive strength(MPa)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>3 d</td>
<td>7 d</td>
</tr>
<tr>
<td>Blank</td>
<td>0</td>
<td>Eligible</td>
<td>31.2</td>
</tr>
<tr>
<td>TEOA</td>
<td>0.03</td>
<td>Eligible</td>
<td>34.3</td>
</tr>
<tr>
<td></td>
<td>0.05</td>
<td>Eligible</td>
<td>32.5</td>
</tr>
<tr>
<td>DGA</td>
<td>0.03</td>
<td>Eligible</td>
<td>36.1</td>
</tr>
<tr>
<td></td>
<td>0.05</td>
<td>Eligible</td>
<td>35.6</td>
</tr>
</tbody>
</table>

**C. Effect of DGA and TEOA on Physical Mechanical Performance of Cement**

Physical and mechanical properties of cement with different types and dosages of grinding aids are shown in Tab. 3.

It can be seen from Tab.3 that soundness of all cement are eligible, the compressive strength of cement with DGA at 3, 7 and 28 d are improved significantly. And the effect is the best when the dosage of DGA is 0.03%. The compressive strengths of cement at 3, 7 and 28 d are respectively increased by 4.9, 5 and 3.5 MPa. When the dosage of DGA is 0.03%(it is 1.8, 1.6 and 4.3 MPa respectively higher than TEOA at the same dosage). On one hand, the enhancement effect is directly related the content of 3~32 μm particles of cement which is improved significantly by adding DGA. On the other hand, the carboxyl of the grinding aids reacted with Ca²⁺ in cement, which could accelerate the hydration process of C₃S, produce insoluble calcium salt and improve the strength and density of hydration product. The complexation of carboxyl groups promoted the hydrolysis of C₃A in the early stage of cement reaction, forming the gelatinous AFt, and then improved the early strength of cement.

**D. SEM Analysis of Cement Hydration Products**

The cement paste micro test was carried out with DGA at the dosage of 0.03%.

SEM of hydration products for cement paste at 3 and 28 d are shown in Fig. 2.

It can be seen from Fig. 2 that hydration product structure of blank cement at 3 d is relatively loose and has more pores. It contains a large amount of cement clinker minerals that it hasn’t occurred hydration reaction. A small amount of layered Ca(OH)₂ is fuzzy visible and Amorphous C-S-H gel is seldom distributed in hydration structure. However, hydration product structure of cement with DGA at 3 d is relatively dense, its structure has a little number of cement clinker minerals that it hasn’t occurred hydration reaction. C-S-H and Ca(OH)₂ in hydration structure overlap with each other form a more stable micro-structure.
It can be seen from Fig.2 that hydration product structure of blank cement at 28 d is have a large number of gelatinous AFt, the morphology of which has been unable to identify and the structure is relatively dense. The hydration of cement paste with DGA is more thoroughly and paste structure is filled fully with hydration product. A large amount of layered Ca(OH)$_2$ is fuzzy visible. The pore of the cement stone is very few, is formed integrally, and then the structure of hydration product is firm.

In any case, the improved microstructure by addition of DGA resulted in the better strength development, and it promoted reaction process of cement hydration, improved hydration degree of cement, and then more hydration products were produced in the hydration process of cement.

E. XRD Analysis of Cement Hydration Products

XRD of hydration products for harden cement paste at 3 and 28 d are shown in Fig. 3 and Fig. 4, respectively.

![Figure 3. XRD of hydration product for cement paste at 3 d](image)

![Figure 4. XRD of hydration product for cement paste at 28 d](image)

It can be seen from Fig. 3 and Fig. 4 that the main hydration products are Ca(OH)$_2$, AFt and C-S-H. Compared with the blank group, the diffraction peaks of Ca(OH)$_2$, AFt and C-S-H in hydration products of cement with DGA at 3 and 28 d are stronger. This is due to hydration speed of cement with DGA is faster, and more hydration products such as Ca(OH)$_2$ is produced. However the diffraction peaks of C$_3$S, C$_2$S are weak, indicating that C$_3$S and C$_2$S are more involved in the hydration reaction.

The addition of DGA improved the strength of cement paste at different ages. On one hand, the addition of DGA reduced the surface energy of cement particles and improved the effective particle size distribution.

On the other hand, DGA could react with Ca$^{2+}$ accelerating the hydration reaction rate, and then produced more hydration products, which is beneficial to the development of later strength.

So it indicates that DGA can promotes hydration reaction of clinker, thus strength of cement is improved.

IV. CONCLUSIONS

- The optimization effect of DGA particles size distribution of cement was better than TEOA. The specific surface area of cement with 0.03% DGA was 14 m$^2$/kg more than that with TEOA at the same dosage. The content of 3~32 μm particles of cement with 0.03% DGA was 1.9% higher than that with TEOA at the same dosage.
- DGA could improve significantly mechanical property of cement. The compressive strength of cement with 0.03% DGA at 3, 7 and 28 d was increased by 1.8, 1.6 and 4.3 MPa, respectively.
- DGA had the amino, hydroxyl and carboxyl groups, which easily combined with metalions of cement particles, resulting in that DGA could promote surface adsorption and shield the unsaturated charges in the surface and crack section of particles, and then particles reunion was prevented. Thus DGA exhibited an excellent dispersing and grinding effect.

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REFERENCES