Review of recent research in nano cellulose preparation and application from jute fibers

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Abstract. Jute is the second most important fiber in the world, which is easy to grow on all types of soil from clay to sandy loam. The popularity of jute is attributable to its abundant and cheapness, and jute fiber is mainly used in the manufacture of coarse fabrics like bags, hessian, sacking, canvas and twines. Jute is rich in cellulose, which could be served as promising candidates for nano materials production. Nano cellulose features concerned its high strength and stiffness, renewability, low density, biodegradability, low thermal expansion coefficient and high aspect ratio. Fields of use of nanocellulose include reinforcement of nanocomposite materials, thickening agents, tissue engineering scaffolds, dry-strength additive for paper, filtration media, adsorbents, etc. This review summarizes progress in nanocellulose preparation and also discusses recent developments in nanocomposite fabrication based on jute fibers.

1. Introduction

Jute is the cheapest and the second most common natural fiber, which is used extensively in the manufacture of different types of packaging material for various agricultural and industrial products. Jute fiber is extracted from the inner bast tissues of the bark of the plant’s stem\textsuperscript{[1]}. Corchorus Capsularis L.(White) and Corchorus.olitorius L.(Tossa), mostly cultivated for fibers, belong to the genus Corchorus, family Tiliaceae. The genus includes about 40 species. Jute is grown extensively in Bangladesh, India and China\textsuperscript{[2]}.

Development of nanomaterial’s based on nanocellulosic substances is a rather new but rapidly developing research area. The nanocellulose exhibit excellent properties like high strength, high stiffness, high aspect ratio, low cost, renewability and biodegradability\textsuperscript{[3]}. Many plants could be used as resources for nanocellulose extraction such as jute\textsuperscript{[4]}, sisal\textsuperscript{[5]}, rice husk\textsuperscript{[6]}. The nanocellulose is suggested a wide range of potential commercial uses such as biomedicine\textsuperscript{[7]}, tissue engineering scaffolds\textsuperscript{[8]}. The use of jute fibers in the nanomaterials area can bring prosperity and economic benefit of the traditional fiber, particularly from an environmental point of view. The aim of this review is to describe current state to the preparation and application of nanocellulosic materials from jute fibers.

2. Structure of Jute fibers

Jute fibers composed mainly of cellulose (59-71%), hemicellulose (12-13%), lignin (11.8-12.9%) , pectin (0.2-4.4%) and wax/fat (0.5\%)\textsuperscript{[9]}. The lignin content of jute was high compared to other bast fibers. Therefore, it is difficulty to extract the cellulose with high purity. Cellulose is composed of β-d- glucopyranose units joined by (1→4) glycosidic links\textsuperscript{[10]}. The hemicellulose of jute fiber is thought to consist mostly of polyuronides hexosans and xylan. Lignins are amorphous polymers consisting of phenylpropane units\textsuperscript{[11]}. The reactivity of hydroxyl groups (-OH) of cellulose at the C-2, C-3 and C-6 positions influenced the physical and chemical properties of the cellulose chain\textsuperscript{[12]}. The interaction between the cellulose chains is tight due to the weaker Van der waals force and stronger intermolecular and intramolecular H-bonding\textsuperscript{[13]}. It is extremely difficult to extract nanocellulose with relatively uniform particle size.

Morphologically, Jute fiber shows the multicellular appearance with hierarchical structure, which can be described as the concentric layers surrounding the lumen, and the surfaces covered
with the gummy matter[14]. Jute fibers is composed of a number of ultimate cells with lignin and hemicellulose acting as cementing agents to give strength and flexibility to the fiber[15]. Several studies suggest that most of the hemicelluloses are present in the inter-fibrillar matrix within the ultimate cells, and most of the lignin is believed to be located in the middle lamella between the ultimate cells[16]. The outermost layer is the primary wall (P), followed by the secondary wall (outer layer S1, middle layer S2, inner layer S3). Jute cell wall organization as shown in Fig. 1. In the primary layer, the elementary fibrils have in general a crisscross arrangement. And the secondary layer straight fibrils are observed in the axial section. Some helically oriented fibrils are present in the upper portion of the secondary layer[17].

![Fig.1 Scheme of the jute cell wall organization][2,9,17]

Cellulose molecules are linked together to form elementary fibrils, which have a diameter of around 2–20 nm depending on their origin. The elementary fibril is made up of amorphous and crystalline parts[18]. The crystalline areas are mainly connected with the rigidity, density and swelling properties in the fiber structure; whereas the less ordered amorphous areas are mainly refers to the fiber softness, flexibility, and extensibility[19]. Some researchers investigated jute microstructure. The lateral width of the microfibrils has been measured as 28.0±3.0Å by Heyn with negative staining of sections methods [20]. Mukherjee extracted elongated particles from jute by sulphuric acid. The mean width and thickness of the particles are about 85Å and 20Å respectively[21].

3. Jute nanocellulose

Two kinds of nanoparticles could be extracted from jute: Microfibrillated Cellulose (MFC) with web like structure and nano cellulose whiskers (NCW) with elongated crystalline rod-like appearance. The NCW is used to designate nanoparticles with a relatively low aspect ratio, whereas the MFC is referred as cellulose microfibrils with lateral dimensions around 10 to 100 nm. The webs like structure and micrometer length scale are the distinct characters of MFC[22].

3.1 Microfibrillated Cellulose (MFC)

Microfibrillated Cellulose (MFC) is obtained through a mechanical treatment such as high pressure homogenizing, grinding, microfluidizer, ultrasonication, ball milling, cryocrushing and steam explosion. Purification by the chemical pretreatment usually was carried out before nanofibrillation for lignocellulosic fibers. This process results in the solubilisation of lignin, pectins and hemicelluloses.

Microfibrillated cellulose(MFC) were extracted by Alila et al.[23] from of several non-woody plants (basts of flax, hemp, jute, leaves of sisal and abaca) under high pressure homogenization process at a pressure lower than 600 bar. The pretreatment of alkaline pulping process and TEMPO-mediated oxidation were assistance for the fibrillation process. The fibrils width of jute...
with web-like structure was in the 30–50 nm range and several microns length. They also found that the fibers with the highest content of hemicelluloses help to get the highest yield in MFC.

Baheti et al. [24] extracted nanofiber from short fibers of jute spinning waste by high energy planetary ball milling process. The particle size was below 500 nm with narrow size distribution after 3 hours of wet milling in the deionized water. Wang et al. [25] prepared micro- and nano-fibrils from jute by chemical and high pressure steam treatments. Micro-fibrils and nano-fibrils are observed and the diameter of the fibrils is about 0.1–3 μm. Abraham et al. [26] obtained an aqueous stable colloid suspension of cellulose nano fibrils from jute using steam explosion technique along with mild chemical treatment. The jute nanocellulose size is in the range of 15–25 nm in diameter. Lin et al. [27] extracted the cellulose nanofibrils from the jute fibers after the pre-treatments followed by the TEMPO mediated oxidation and mechanical disintegration. The diameter of cellulose nanofibrils was at a range from 5 nm to 20 nm and a length of several micrometers. Cellulose aerogels was formed by the lyophilization of nanofibrils dispersions with different concentrations. The cellulose aerogels has many potential applications like pharmaceutical, liquid filtration and tissue engineering scaffolds.

3.2 Nano cellulose whiskers

Jute cellulose nanowhiskers (NCW) are a kind of renewable and biocompatible nanomaterials cause much interest because of its versatility in various applications. NCW exhibited elastic moduli as high as 100–150 GPa, and high aspect ratio (10–100), with diameters ranging from 4 to 20 nm [28].

The elementary unit of jute cellulose I structure is monoclinic. Cellulose is a carbohydrate of very high molecular weight and represented by the formula \((\text{C}_6\text{H}_{10}\text{O}_5)\text{n}\). These elementary units are repeated at regular spacing along the length of the fiber to form one long fiber molecule. One micelle is formed by such several molecular fibers in sideways arrangement as in a crystal. These micelles are arranged parallel to each other along the chain length in fibers [29]. The cellulose structure is divided into two regions, a crystalline region and an amorphous region, which is responsible for the transverse cleavage of the microfibrils into short monocrystals under acid hydrolysis [30]. Among the commercially available acids, the commonly used are hydrochloric, sulphuric, formic and phosphoric acid. The most frequently used is sulphuric acid. Due to the \(\text{H}_2\text{SO}_4\) prepared whiskers present a negatively charged surface, which leads to more table whiskers aqueous suspension than that prepared using hydrochloric acid [31]. The oxidation of the whiskers or the post-sulfation of HCl is the other options to get charged whisker [32,33].

Kasyapi et al. [34] extracted bionanowhiskers from jute by acid hydrolysis combined with alkali pretreatment. After 1h of acid hydrolysis, the rod like morphology was observed. And the nano whiskers with the length and width of 550 ± 100 nm, 77±30nm respectively. These nanowhiskers might be useful as reinforcing phase in nanocomposites. Jahan et al. [35] extracted cellulose from jute fibers at a yield of 59.8% based on the formic acid & peroxy formic acid process at an atmospheric pressure. The microcrystalline cellulose were further to be obtained by sulphuric acid hydrolysis process. The diameter of nanofibers was 15–40 nm. Cao et al. [4] extracted cellulose nanowhiskers from jute fibers by a 2,2,6,6-tetramethylpiperidine -1-oxyl radical (TEMPO)/ NaBr/ NaClO system selective oxidation combined with mechanical homogenization with over 80% yield. The diameters of jute cellulose nanowhiskers was around 3–10 nm with high crystallinity (69.72%). Cao et al. [36] also obtained spider-web-like nanoporous networks by simple directly immersion-drying method. The jute cellulose nanowhiskers (3–10 nm) deposited on the electrospun nanofibrous membrane (100–300nm), formed the spider-web-like nanoporous networks, which has potential useful for filter applications.

4. Jute nanocomposites

Nanocomposites in general are two-phase materials, and one of the phases has at least one dimension in the nanometer range (1–100 nm). The advantages of nanocomposite materials when compared with traditional composites are their superior thermal, mechanical, barrier properties, recyclability, transparency and low weight even at low volume fraction of nanofibers [37].

The first report of the use of cellulose nanofibers as a reinforcing phase in composite materials was by Favier et al. [38] in 1995. Since then a huge number of literature has been related to cellulose
nano composite, and it is becoming an increasingly popular subject. Many methods about nano cellulose surface modification have been applied to improve the compatibility between nanofibers with matrices such as corona, plasma discharges, graft copolymerization or application of surfactant[39].

Kunal Das et al.[40,41] prepared nanofibers from jute by acid hydrolysis method. Starch/polyvinyl alcohol (PVA) & Biocopolyester matrix based biocomposite reinforced with jute nanofibers at different loading (5, 10 and 15 wt. %). The 10 wt. % jute nanofibers loaded films exhibited best combination of properties. Baheti prepared nanofibers for reinforcement of polylactic acid (PLA) films by solvent casting. The nanofibers extracted from jute waste fibers with size around 50~500 nm after by high energy planetary ball milling in wet condition for 3 hours. After added of 5 wt. % jute nanoparticles, it was observed that glass transition temperature (Tg) of PVA films improved from 84.36°C to 95.22°C. Dynamic mechanical analysis of composite films showed higher value of 14×10^8 Pa for storage modulus in comparison to 9×10^8Pa of neat composite film[42]. The Young’s modulus was increased to 3.3GPa from 1.0GPa as compared with neat PLA film[43,44]. The initial modulus and tensile strength increased by 217.30% and 170.59%, respectively as compared to neat PLA film[45]. The mechanical properties of 3% loading of treated jute nanofibrils as reinforcement in PLA matrix are also found to be improved. Comparing with the pure PLA film, the initial modulus and tensile strength increased by 207.69% and 168.67% respectively[46]. The composite film has the potential applications such as food packaging, agriculture mulch films, automotive plastics, etc. Padal used jute nanofibers as reinforcement in epoxy resin matrix composites. The composites were prepared 50wt. % of glass fiber and 50wt. % epoxy resin content. Cellulose nanofibers are extracted from jute by a chemical and high energy ball milling. The nano particle size were distributed from micro to 20~50 nm with spherical and elliptical shape. The mechanical properties of nanocomposite has improve remarkably when 3 wt. % Jute was added as reinforcement[47]. The tensile strength of nanocomposite was increased to 96% with the 3wt. % nanofiber reinforcement as compared to the pure composite. This composite has the potential for high damping applications because the damping parameters of the nanofiber composites much higher than the pure composites[48]. Maiti et al.[49] prepared biocopolyester composite sheets with 2% and 10 wt. % Jute nano fiber (JNF), and compostability tests were performed in simulated aerobic compost environment at ambient temperature for a period of 50 days. Weight loss study revealed that the incorporation of JNF enhanced the rate of degradation significantly. Molecular weight study revealed the fact that biocopolyester molecules had a significant breakdown in chain length during melt mixing with 10 wt. % JNF. Rahman et al.[50] extracted crystalline cellulose from jute by hydrolysis with 40% H2SO4 to get mixture of micro/nanocrystals. Biodegradable nanocomposites were prepared by extrusion and hot press method using jute crystalline cellulose (3-15%) and poly (lactic acid). Vicker hardness and yield strength were found to increase with increasing cellulose content in the composite. The antibacterial effect were found after filling with 15% crystalline cellulose in the composite. The composite might be suitable for biomedical application because of the antibacterial effect and its cytotoxicity.

5. Conclusions

This article summarizes some of the recent advances in extraction of nanofibers from jute and their potential uses in nanocomposite. Jute fiber is a cheap, abundant, and renewable source with high cellulose content like 60~70%. It’s found to be excellent precursor for the extraction of cellulose nanoparticles. Owing to its outstanding performance, the nanofiber of jute has already become an important and hot spot in the nanomaterials study. Furthermore, application of jute nanofibers as reinforcement improves polymer mechanical properties such as tensile strength and modulus even at very low volume fraction. The nanocomposite could be used as packaging, electronic device, biomedical, cosmetic, tissue engineering scaffolds, filtration media, rheology modifiers, adsorbents, paper reinforcement and so on. As lignocellulosic fiber, there are some problems need to be solved in order to nanofiber commercial production: (1) The removal of noncellulosic materials of fibers with high efficiency is helpful for nanoparticle size reduction and nanofiber-matrix adhesion. (2) The nanofiber disintegration process should be optimized for the reduction of high energy consumption. (3) The requirements of chemical modification of nanofibers
in order to improve compatibility with apolar matrix. The reactive hydroxyl groups on the surface of nano cellulose also provide the possibility for fabricating a wide range of functionalized materials for future advanced applications.

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